

#### **Sodium Metabisulfite**

# SECTION 1: IDENTIFICATION OF THE SUBSTANCE/MIXTURE AND OF THE **COMPANY/UNDERTAKING**Product identification:

### **Product Description:**

Sodium metabisulfite

CAS-No: 7681-57-4 **EC-No.:** 231-673-0

Molecular Formula: Na<sub>2</sub>S<sub>2</sub>O<sub>5</sub>

Relevant identified uses of the substance or mixture and uses advised against:

**Details of the supplier of the safety data sheet:** 

**Company** Sulfurshimi Kashan

No. 315, Kolahdooz St, Pasdaran

Ave, Tehran, Iran

Web: www.sulfurshimi.com

E-mail Address info@sulfurshimi.com

### **Emergency telephone number:**

For Emergency contact on: +98 - 2122 - 765743

#### **SECTION 2: HAZARDS IDENTIFICATION**

**Warning Statement:** None

**Hazard Rating:** Health=2 Fire=0 PPE\_Sec=8 Reactivity=0

**Primary Entry Routes:** Inhalation

**Target Organs** Respiratory system. Eyes. Skin

**Acute Effects** Acute effects to exposure of sodium metabisulfite includes eye and

mucous membrane irritation. Decomposition of Sodium Metabisulfite (and solutions) may release toxic and hazardous fumes of sulfur oxides, including sulfur dioxide. Acute poisoning from sulfur dioxide is rare because the gas is easily detected. It is so irritating that contact cannot be tolerated. Symptoms include coughing, hoarseness, sneezing, tearing, and breathing difficulty. However, workers who cannot escape exposure may suffer severe pulmonary damage which can be fatal.

Irritant

Inhalation



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Eye Irritant Skin Irritant

**Ingestion** Not likely to occur

Carcinogenicity IARC, NTP, and OSHA do not list sodium metabisulfite as a

carcinogen

Chronic Effects Prolonged or repeated exposure may cause dermatitis, and sensitization

reactions.

Medical Conditions Aggravated by Long term Exposure

Capable of provoking bronchospasm in sulfite

sensitive individuals who have asthma.

### **SECTION 3: COMPOSITION/INFORMATION ON INGREDIENTS**

**Substances:** Sodium Metabisulfite

**Mixtures:** 

Component	TWA	CAS-No	EC-No.	Weight %
Sodium Metabisulfite	5 mg/m <sup>3</sup>	7681-57-4	231-673-0	98

### **SECTION 4: FIRST AID MEASURES**

**Inhalation** sore throat, shortness of breath Remove from exposure to

fresh air. Seek coughing, and congestion.

Eye Contac Irritation to eyes and mucous membranes. Irrigate with water

until no evidence of chemical remains. Obtain medical

attention.

**Skin contact** Irrigation, itching, dermatitis. Wash with soap and drench

With water. Remove contaminated clothing and wash before

reuse.

**Ingestion** Irritation to mucous membranes. Give large amount of water

or milk. Immediately obtain medical attention.

After first aid, get appropriate medical attention.



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## **SECTION 5: FIREFIGHTING MEASURES**

Flash point

Flash Point method

Not Applicable
Burning rate

Auto ignition

LEL

Not Applicable

Flammability

**Extinguishing media**Use extinguishing agent appropriate for surrounding fire

conditions.

**Hazardous combustion products** may release hazardous gas.

**Fire-fighting instructions** do not release runoff from fire control methods to sewers

or water ways.

Fire-fighting Equipment because fire may produce toxic thermal decomposition

products, wear a self-contained breathing apparatus

(SCBA) with a full face piece operated in pressure-demand

or positive-pressure mode.

## **SECTION 6: ACCIDENTAL RELEASE MEASURES**

**Spill/leak procedures** wear appropriate PPE according to section 8.

Small Spills/leaks Spills can be neutralized with an alkaline material such as

caustic soda. Leaks may be located by spraying the area with ammonium hydroxide solution which forms a white fume in

the presence of sulfur dioxide.

Large spills/leaks large spills should be handled according to a predetermined

plan.

**Containment** for large spills, dike far ahead of contaminated runoff for later

## **SECTION 7: HANDLING AND STORAGE**

### Precautions for safe handling

Avoid contact with product. Do not breathe dust or vapor.

### Conditions for safe storage, including any incompatibilities:

Avoid heat or moisture. Store in areas, away from heat and moisture and protected from physical damage. Segregate from acids and oxidizers.



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## SECTION 8: EXPOSURE CONTROLS/PERSONAL PROTECTION

**Ventilation** Provide general or local exhaust ventilation systems to

maintain airborne concentrations below OSHA limits. Local

exhaust ventilation is preferred because it prevents

contaminant dispersion into the work area by controlling it at

the source.

**Respiratory protection** Follow OSHA respirator regulations (29 CFR 1910.134) and,

If necessary, wear a MSHA/NIOSH-approved respirator. Select respirator based on its suitability to provide adequate worker protection for given working conditions, level of airborne contamination, and presence of sufficient oxygen. For emergency or on-routine operations (cleaning spills, reactor vessels, or storage tanks), wear a SCBA. Warning! Air-purifying respirators do not protect workers in oxygen-

deficient atmospheres.

Protective clothing/equipment Wear protective gloves, boots, and clothing when necessary

to prevent excessive skin contact. Wear protective eyeglasses or goggles, per OSHA eye-and face-protection regulations

(29 CFR 1910.133).

Safety stations Make emergency eyewash stations, showers, and washing

facilities available in the work area.

Contaminated equipment Remove this material from personal protective rquipment as

needed.

**Comments** Do not eat, drink, or smoke in work areas. Practice good

personal hygiene after using this material, especially before

food or beverage consumption.

## **SECTION 9: PHYSICAL AND CHEMICAL PROPERTIES**

### Information on basic physical and chemical properties:

• **Appearance:** White

• Physical State: Solid crystal

• **Odor Threshold:** pungent SO<sub>2</sub> odor

• pH: Acidic

Melting Point: 150°C

Molecular weight: 190.11

Specific gravity ( $H_2O=1$ ): 1.5

• Water solubility: 45% @ 200 ° C



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## **SECTION 10: STABILITY AND REACTIVITY**

Stability Stable under normal conditions.

**Polymerization** Hazardous polymerization will not occur.

**Chemical Incompatibilities** 

In the presence of water, or acid, sodium Metabisulfite (and solutions) may release toxic and hazardous fumes of sulfur oxides, including sulfur dioxide. Acute poisoning from sulfur dioxide is rare because the gas is easily detected. It is so irritating that contact cannot be tolerated. Symptoms include coughing, hoarseness, sneezing, tearing, and breathing difficulty, however, workers who cannot escape high accidental exposure may suffer severe pulmonary damage which can be fatal. Contact with powdered potassium, sodium metals, alkali, and oxidizing agents produce violent reactions. Reacts with water and steam to form corrosive sulfurous acid. Reacts with chlorates to form unstable chlorine dioxide.

**Conditions to avoid** Avoid excessive heat, or open flame, and moisture.

**Hazardous decomposition products**May release hazardous sulfur dioxide gas.

## **SECTION 11: TOXICOLOGICAL INFORMATION**

Eye effects (rabbit)Not availableSkin effects (rabbit)Non-CorrosiveAcute Inhalation effects (rat)Not availableAcute Oral effect (rat)LD50=115 mg/kg

Carcinogenicity IARC, NTP, and OSHA do not list sodium metabisulfite as a

carcinogen.

Chronic effects

Prolonged or repeated exposure may cause dermatitis, and sensitization reactions. Exposure to asthmatic, atopic and sulfite sensitive individuals may result in severe bronchoconstriction and reduced levels in forced expiratory volume. Decomposition of sodium metabisulfite (and solutions) may release toxic and hazardous fumes of sulfur

solutions) may release toxic and hazardous fumes of sulfur oxides, including sulfur dioxide, which may cause permanent pulmonary impairments from acute and chronic exposure. The immediately dangerous to life or health (IDLH) level for SO<sub>2</sub>

is 100 ppm.

### **SECTION 12: ECOLOGICAL INFORMATION**

**Eco toxicity** sodium metabisulfite is a non-hazardous solid

commonly used as a waste water dechlorinating agent. High concentrations will contribute to elevated chemical oxygen

demand in aquatic environments.

**Environmental transport** Soluble in water

**Environmental Degradation** Rapid biological decomposition.



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Soil Absorption/Mobility Slight.

## **SECTION 13: DISPOSAL CONSIDERATIONS**

**Disposal** waste determinations typically consider sodium metabisulfite

contaminated materials to be non-hazardous.

**Disposal regulatory requirements**Follow applicable federal, state and local regulations. **Container Cleaning and disposal**Follow applicable federal, state and local regulations.

### **SECTION 14: TRANSPORT INFORMATION**

### **DOT Transportation Data (49 CFR 172.101)**

Shipping Name Sodium Metabisulfite, non-regulated material

Shipping symbols NA
Hazard Class NA
Subsidiary Hazard NA
ID No. NA
Packing Group NA
Label NA
Special Provisions NA

### **SECTION 15: REGULATORY INFORMATION**

**EPA Regulations:** 

RCRA Hazardous Waste Classification (40 CFR 261)

RCRA Hazardous Waste Number (40 CFR 261)

Not listed
CERCLA Hazardous substance (40 CFR 302.4)

CERCLA Reportable Quantity (RQ)

NA

SARA Title III. goetien 202 Entroppely Hazardous Substance

SARA Title |||: section 302 Extremely Hazardous Substance: Not listed FIFRA Not regulated

**OSHA Regulations:** 

Air Contaminant (29 CFR 1910.1000)

OSHA Specifically regulated substance

Not listed

**Other Regulations:** 

**FDA** Regulated when used as food preservative.

### **SECTION 16: OTHER INFORMATION**

The information herein is believed to be reliable. However, no warranty, expressed or implied, is made as to its accuracy or completeness and none is made as to the fitness of this material for any purpose. The manufacturer shall not be liable for damages to person or property resulting from its use. Nothing herein shall be construed as a recommendation for use in violation of any patent.